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## Key indicators

Single-crystal X-ray study
$T=193 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.059$
$w R$ factor $=0.120$
Data-to-parameter ratio $=10.5$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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# Ethyl 3-(4-geranyloxy-3-methoxyphenyl)-2-propenoate 

The title compound, $\mathrm{C}_{22} \mathrm{H}_{30} \mathrm{O}_{4}$, is known as a chemopreventive agent against colon tumor development. The molecular structure is nearly planar, showing a trans arrangement with respect to the $\mathrm{C}=\mathrm{C}$ bond of the phenylpropenoate moiety.

## Comment

The title compound, (I), is known as a chemopreventive agent against colon tumor development (Tsuda et al., 1999). It was prepared from ferulic acid, which was easily prepared in large quantities from the oily component of rice bran (Taniguchi et al., 1996).


EGMP

The molecular structure of (I) is shown in Fig. 1, confirming the trans arrangement with respect to the $\mathrm{C} 7=\mathrm{C} 8$ double bond, similar to ferulic acid (Nethaji et al., 1988). The orientations of the phenyl ring and the geranyl group are defined by the torsion angles in Table 1. The molecule is nearly planar. The molecules in the crystal lattice are in a conventional herring-bone-type arrangement (Fig. 2). No close contacts were found that indicated any relevant intermolecular interaction.

## Experimental

To a solution of $1.0 \mathrm{~g}(4.5 \mathrm{mmol})$ of ethyl ferulate, ethyl ester of ferulic acid, $1.2 \mathrm{~g}(4.6 \mathrm{mmol})$ of triphenylphosphine and 0.69 g ( 4.5 mmol ) of geraniol in 10 ml anhydrous tetrahydrofuran was added $0.93 \mathrm{~g}(4.6 \mathrm{mmol})$ diisopropyl azodicarboxylate. The mixture was stirred for 5 min at room temperature and the solvent then removed under reduced pressure. After the residue was extracted with 30 ml of hexane, the hexane was removed under reduced pressure. Recrystallization from methanol gave the pure title compound (I) $(1.37 \mathrm{~g}, 85 \%)$. Crystals of (I) were obtained by slow evaporation of an ethanol solution at 277-279 K.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{22} \mathrm{H}_{30} \mathrm{O}_{4} \\
& M_{r}=358.48 \\
& \text { Triclinic, } P \overline{1} \\
& a=11.094(1) \AA \\
& b=11.943(1) \AA \\
& c=9.3399(9) \AA \\
& \alpha=110.864(5)^{\circ} \\
& \beta=113.510(5)^{\circ} \\
& \gamma=68.358(4)^{\circ} \\
& V=1022.5(2) \AA^{\circ}
\end{aligned}
$$

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Figure 1
The molecular structure of (I) showing $50 \%$ probability displacement ellipsoids.

## Data collection

Rigaku R-AXIS-RAPID imageplate diffractometer
$\omega$ scans
Absorption correction: multi-scan (Higashi, 1995)
$T_{\text {min }}=0.587, T_{\text {max }}=0.828$
5896 measured reflections

## Refinement

Refinement on $F$
$R=0.059$
$w R=0.121$
$S=1.65$
3429 reflections
326 parameters
H -atom parameters constrained

3429 independent reflections
3094 reflections with $F^{2}>2.0 \sigma\left(F^{2}\right)$
$R_{\text {int }}=0.028$
$\theta_{\text {max }}=68.2^{\circ}$
$h=-13 \rightarrow 13$
$k=-14 \rightarrow 14$
$l=-11 \rightarrow 11$
$w=1 /\left[\sigma^{2}\left(F_{o}\right)+0.00403\left|F_{o}\right|^{2}\right]$
$(\Delta / \sigma)_{\max }<0.001$
$\Delta \rho_{\text {max }}=0.51 \mathrm{e}^{\AA^{-3}}$
$\Delta \rho_{\text {min }}=-0.28$ e $\AA^{-3}$
Extinction correction: Zachariasen (1967) type 2 Gaussian isotropic Extinction coefficient: 0.39 (6)

Table 1
Selected torsion angles $\left({ }^{\circ}\right)$.

| $\mathrm{O}(1)-\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)$ | $180.0(1)$ |
| :--- | ---: |
| $\mathrm{O}(3)-\mathrm{C}(9)-\mathrm{O}(4)-\mathrm{C}(10)$ | $-1.1(3)$ |
| $\mathrm{O}(3)-\mathrm{C}(9)-\mathrm{C}(8)-\mathrm{C}(7)$ | $-17.8(2)$ |
| $\mathrm{O}(4)-\mathrm{C}(9)-\mathrm{C}(8)-\mathrm{C}(7)$ | $3.2(3)$ |
| $\mathrm{C}(1)-\mathrm{O}(1)-\mathrm{C}(13)-\mathrm{C}(14)$ | $-178.2(1)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{O}(2)-\mathrm{C}(12)$ | $177.6(2)$ |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{O}(1)-\mathrm{C}(13)$ | $179.4(1)$ |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{O}(2)-\mathrm{C}(12)$ | $-3.6(3)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(7)-\mathrm{C}(8)$ | $174.6(2)$ |
| $\mathrm{C}(4)-\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | $178.4(2)$ |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(7)-\mathrm{C}(8)$ | $-5.9(3)$ |
| $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{O}(1)-\mathrm{C}(13)$ | $0.6(2)$ |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{O}(4)-\mathrm{C}(10)$ | $178.9(1)$ |
| $\mathrm{C}(9)-\mathrm{O}(4)-\mathrm{C}(10)-\mathrm{C}(11)$ | $-177.5(2)$ |
| $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)$ | $178.3(1)$ |
| $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(21)$ | $-0.1(3)$ |
| $\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)$ | $-2.3(2)$ |
| $\mathrm{C}(15)-\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)$ | $-177.1(1)$ |
| $\mathrm{C}(16)-\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)$ | $170.5(2)$ |
| $\mathrm{C}(17)-\mathrm{C}(16)-\mathrm{C}(15)-\mathrm{C}(21)$ | $176.1(1)$ |
| $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(20)$ | $179.6(2)$ |
| $\mathrm{C}(17)-\mathrm{C}(18)-\mathrm{C}(19)-\mathrm{C}(22)$ | $0.8(3)$ |

The measured fraction $\left(\theta_{\max }=68.2^{\circ}\right)$ of 0.917 is relatively low, which may be ascribed to either a short exposure time of the image plate or the instrument having only one rotation axis, leaving some parts of reciprocal space inaccessible.

Data collection: PROCESS-AUTO (Rigaku, 1998); cell refinement: PROCESS-AUTO; data reduction: TEXSAN (Molecular Structure Corporation, 1999); program(s) used to solve structure:


Figure 2
Projection down the $c$ axis of the unit cell of (I)

SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: TEXSAN; software used to prepare material for publication: TEXSAN.

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